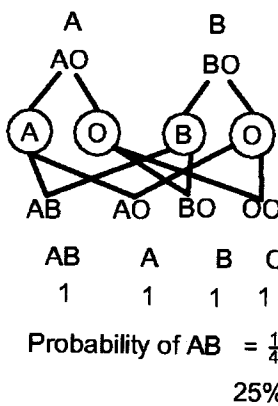
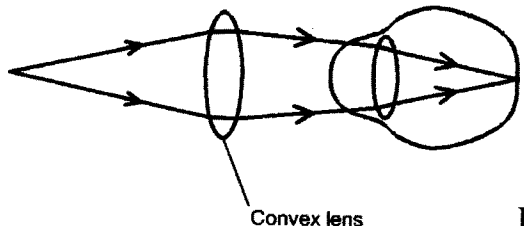


5.6.2 General Science Paper 2 (237/2)

SECTION A: BIOLOGY (34 marks)

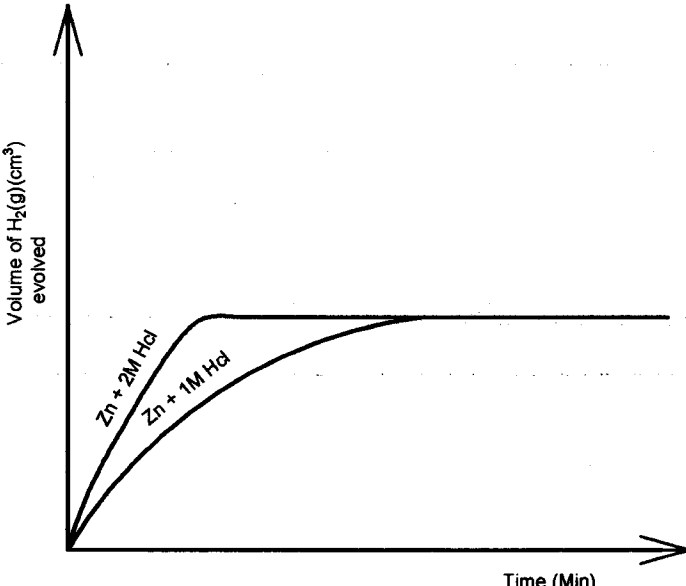
1.	<p>a) - Leaves modified into thorns/are needle like to reduce evapotranspiration;</p> <ul style="list-style-type: none"> - Succulent stems to store water; - Deep rooted to absorb water from deep horizons; - Few stomata to reduce transpiration; - Sunken stomata to reduce rate of transpiration; - Have waxy cuticle to reduce rate of respiration; 	3x1	(3 marks)
	<p>b) - Suffocation of aquatic animals due to reduced dissolved oxygen levels;</p> <ul style="list-style-type: none"> - Rapid growth of aquatic weeds; - Disease agents in sewage contaminated water; - Toxic chemicals in sewage kill aquatic organisms; - Accumulation of heavy metals causing blood poisoning, nervous problems and death; - Rj pollutes the water alone. 	3x1	(3 marks)
2	a) The production of an offspring as a result of the fusion of a male and a female gamete;	1x1	(1 mark)
	b) Is the duration between fertilization and birth;	1x1	(1 mark)
3.	a) Progesterone; RJ wrong spelling.		(1 mark)
	b) Repair and healing of the endometrium/ uterine wall;		(1 mark)
	c) Luteinising hormone (L.H); Rj abbreviations		(1 mark)
	d) Sperms/ova remain viable in the female reproductive system for 2-3 days;		(1 mark)
4.	a) Epigeal germination; RJ wrong spelling.		(1 mark)
	b) Due to faster elongation of the hypocotyl than the epicotyl the cotyledons and plumule are pushed above the ground;		(1 mark)
	c) Protects the plumule/Photosynthesis;		(1 mark)
	d) Exposure of the hypocotyl to light makes auxins to migrate to the lower side; Higher concentration of auxins on the lower side makes the lower sides to grow faster enabling the seedling to straighten;	2x1	(2 marks)
5.	Man/male has X and Y sex chromosomes while woman/female has X chromosomes only; Fusion of X and Y chromosomes produces a male child; while X and X produces a female child;	3 x 1	(3 marks)

6.	<p>(a) Blood group A. Rj AO;AA</p> <p>(b) Blood groups</p>  <p> $\begin{array}{cccc} & AB & A & B & O \\ & 1 & 1 & 1 & 1 \end{array}$ </p> <p>Probability of AB = $\frac{1}{4}$ 25%</p> <p> $\begin{array}{l} F_1 \text{ Genotypes} \\ F_1 \text{ Blood groups} \end{array} \begin{array}{cccc} AB & AO & BO & OO \\ AB & A & B & O; \\ 1 & 1 & 1 & 1 \end{array}$ </p> <p>Probability of AB = $\frac{1}{4}$ / 25%; Rj 0.25</p>	<p>(1 mark)</p> <p>(2 marks)</p> <p>2x1</p>
7.	<p>(a) Hydrotropism; Rj wrong spelling. Acc. Positive hydrotrphism.</p> <p>(b) Enables plant roots to seek for water;</p> <p>(c) On sensing moisture, the auxins migrate to the side with moisture where they inhibit the rate of growth; side away from moisture grows at a higher rate making the root to curve towards the direction of water;</p>	<p>(1 mark)</p> <p>(1 mark)</p> <p>(1 mark)</p>
8.	<p>(a) Long sightedness (hypermetropia). Acc long sighted.</p> <p>(b)</p>  <p>Convex lens</p> <p>D-1 L-1</p>	<p>(1 mark)</p> <p>(2 mark)</p>
9.	<p>(a) Analogous structures</p> <p>(b) Tendrils in passion are modified axillary buds while in pea plant they are modified leaflets but they all carry out the function for providing mechanical support;</p>	<p>(1 mark)</p> <p>(1 mark)</p>
10.	<p>(a) H – Fibula; J – Tibia;</p> <p>(b) Hinge joint;</p>	<p>(1 mark)</p> <p>(1 mark)</p> <p>(1 mark)</p>

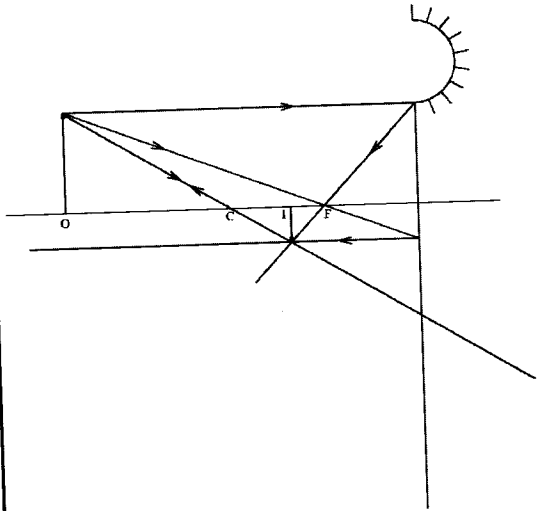
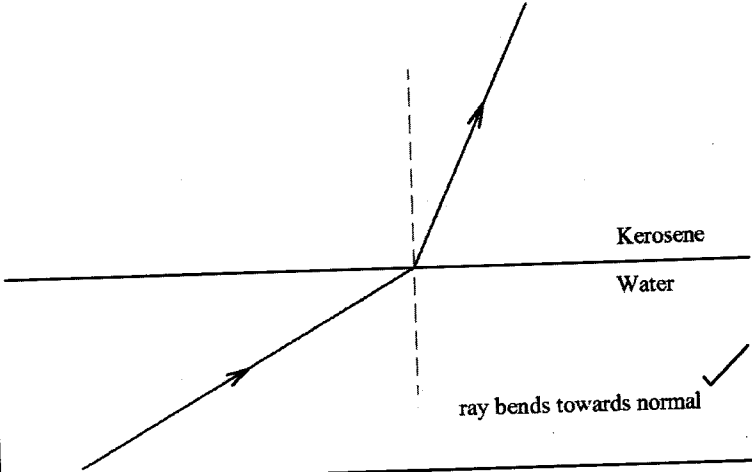
SECTION B - CHEMISTRY


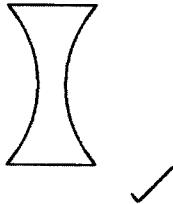
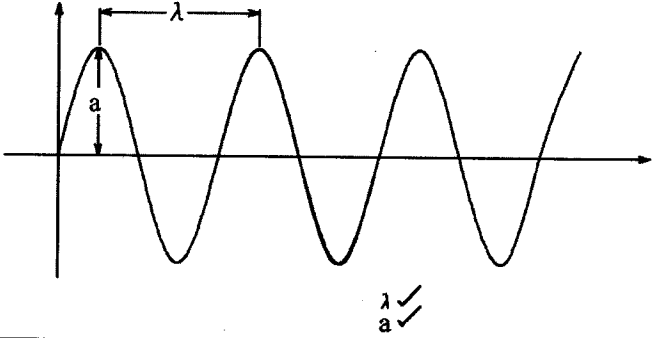
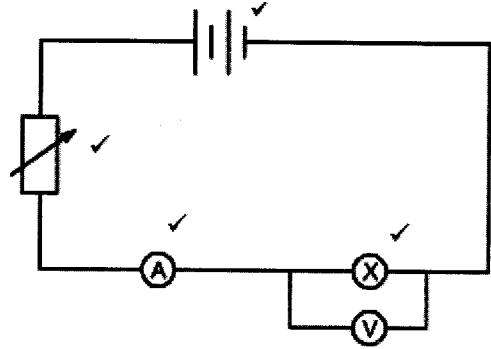
11.	<p>(a)</p> $ \begin{array}{cccc} & \text{H} & \text{H} & \text{H} & \text{H} \\ & & & & \\ \text{H} & -\text{C} & -\text{C} & -\text{C} & -\text{C}-\text{H} \\ & & & & \\ & \text{H} & \text{H} & \text{H} & \text{H} \end{array} $ <p style="text-align: right;">✓1</p>	(1 mark)
	<p>b) manufacture of margarine from oil or manufacture of fats from oils/ _____ Conversion of oils into fats</p>	<p>(1 mark) (2 marks)</p>
12.	<p>(a) $3\text{CuO}_{(s)} + 2\text{NH}_{3(g)} \rightarrow 3\text{Cu}_{(s)} + 3\text{H}_2\text{O}_{(l)} + \text{N}_{2(g)}$</p> <p>(b) reducing agent</p>	<p>(1 mark) (1 mark) 2 marks</p>
13.	<p>a) A mole refers to the amount of a substance that contains as many particles as carbon – 12 or The amount of a substance that contains Avogadro's number of particles. The particles could be atoms, molecules, ions, electrons etc</p>	(1 mark)
	<p>b) $P_2V_2 = P_1V_1$</p> $\Rightarrow p_2 = \frac{p_1v_1}{v_2}$ $= \frac{760 \times 300}{800}$ $= 285\text{mmHg}$	<p>(1 mark) (1 mark)</p>
14.	<p>a)</p>	(3 marks)

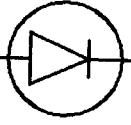
	b(i) Concentrated sulphuric (VI) acid is hygroscopic hence absorbs water from the atmosphere thereby increasing the volume	(1 mark)
	(ii) Used as a drying agent	(1 mark)
15.	(a) The equilibrium will shift to the right as more of the reactants are consumed to produce more RO_3	1 mark
	(b) The yield of RO_3 will decrease because increase in temperature will favour a reaction that absorbs the excess heat.	1 mark
		2 marks
16.	a) (i) ΔH - Heat of solution (ii) Negative (-ve)	(1 mark) (1 mark)
	b) This is because strong acids and strong bases are all fully dissociated/ionized	(1 mark)
	c) - Ease of combustion	(1 mark)
	- Availability	(1 mark)
	- Environmental effects	
	- High heat value	
		(5 marks)
17.	a) Polymer is a big molecule of high molecular mass formed from the reaction of the small molecules.	(1 mark)
	b) - making PVC pipes - Raincoats, handbags, hosepipes, floor tiles	(1 mark)
18.	a) (i) Electrolysis	(1 mark)
	(ii) Reduction	(1 mark)
	b) (i)	(5 marks)
19.	a) Concentration refers to the amount of substance dissolved in a definite volume of solution	(1 mark)

	<p>b (i) No. of moles = $\frac{\text{mass}}{\text{RFM}}$ RFM of NaOH</p> $= 23 + 16 + 1$ $= 40$ $= \frac{28}{40}$ $= 0.7$	<p>(½ marks) 1 mark (½ marks)</p>
	<p>b(ii) 0.7 moles → 2 litres ? → 1 litre</p> $\frac{1}{2} \times 0.7 (\frac{1}{2} \text{ mark}) = 0.35 \text{ M } (\frac{1}{2} \text{ mark})$	
	<p>(iii) 1 mole → 40g 28g → 2l 0.35 moles → ? or ? → 1l</p> $\frac{0.35}{1} \times 40 (\frac{1}{2}) = 14 \text{ g/l } (\frac{1}{2})$ $\frac{1}{2} \times 28 = 14 \text{ g/l}$	5 marks
20.	a) The rate of hydrogen production doubles/ increases	(1 mark)
	<p>b)</p>  <p>The graph shows the volume of hydrogen gas evolved over time for two reactions: Zn + 2M HCl and Zn + 1M HCl. The y-axis is labeled 'Volume of H₂(g) (cm³) evolved' and the x-axis is labeled 'Time (Min)'. Both curves start at the origin and level off to the same final volume. The curve for Zn + 2M HCl rises more steeply, reaching the final volume in less time than the curve for Zn + 1M HCl.</p>	(2 marks)

SECTION C- PHYSICS

21.	<p>(a) Virtual ✓ Upright ✓ Inverted Same size as object</p> <p style="text-align: right;">(Any two correct)</p>	(2 marks)
	<p>(b)</p>  <p>Any two correct rays ✓ 2 Correct position of object(beyond C) ✓</p>	(3 marks)
22.	 <p style="text-align: center;">Kerosene Water</p> <p style="text-align: center;">ray bends towards normal ✓</p>	(1 mark)
23.	<p>(a) Y is positively charged. ✓ X is negatively charged.</p>	(1 mark)
	<p>(b) - Both hair and comb get charged by rubbing. ✓ - Hair attracts air ions of opposite charge and a spark is formed as hair is discharged. ✓</p>	(2 marks)

24.	<p>(a)</p>  <p>Convex lens ✓</p>	(1 mark)
	<p>(b)</p>  <p>Concave lens ✓</p>	(1 mark)
25	 <p>λ ✓ a ✓</p>	(2 marks)
26.	<p>(a) An echo is a second sound heard after the actual (first) sound is reflected from ✓ a barrier./ Reflected sound</p>	(1 marks)
	<p>(b) Wind may increase the speed of sound if in the same direction as the sound. ✓ or reduce the speed of sound if in the opposite direction. ✓</p>	(2 marks)
27.	<p>(a) It aligns itself with the earth's magnetic field. ✓</p>	(1 mark)
	<p>(b) Removal of small pieces of iron from a patient's eye. ✓/electric bells for emergencies</p>	(1 mark)
28.		(4 marks)
29.	<ul style="list-style-type: none"> - The amount of water. ✓ - The power rating of the heater. ✓/amount of current/voltage/resistance/ amount of heat - Pressure of the surrounding. 	(2 marks)

30.	<p>20 → 10 → 5</p> <p>2 half lives in 48 hours ✓ OR</p> <p>∴ half-life = $\frac{48}{2}$</p> <p>= 24 hours ✓</p>	(2 marks)
	$N = N_0 \left(\frac{1}{2}\right)^{t_{1/2}}$ $5 = 20 \left(\frac{1}{2}\right)^{t_{1/2}}$ $t_{1/2} = 24 \text{ hours}$	
31.	<p>(a) The process of adding impurities to a semiconductor in order to increase its conductivity. ✓</p> <p>(b)</p>  <p>✓</p>	(1 mark)
32.	<p>(a) - Carry negative charge. ✓</p> <p>- Deflected by both magnetic and electric fields. ✓</p> <p>- Move in straight lines.</p> <p>- Affect photographic plates.</p> <p>- Cause fluorescence.</p>	(2 marks)
	<p>(b) - X-ray photography ✓</p> <p>- Sterilize equipment</p> <p>- Detect flaws in metals</p> <p style="text-align: right;">(any one)</p>	(1 mark)
33.	<p>- to minimize cost of electricity/ wastage of power ✓</p> <p>- for bulbs to last longer</p> <p style="text-align: right;">(any one)</p>	(1 mark)